

Bodyplethysmography physics

by: Johan Brouwer

R_{aw} images: Dr. H. Eschenbacher

Ideal gas law: $P.V = n.R.T$

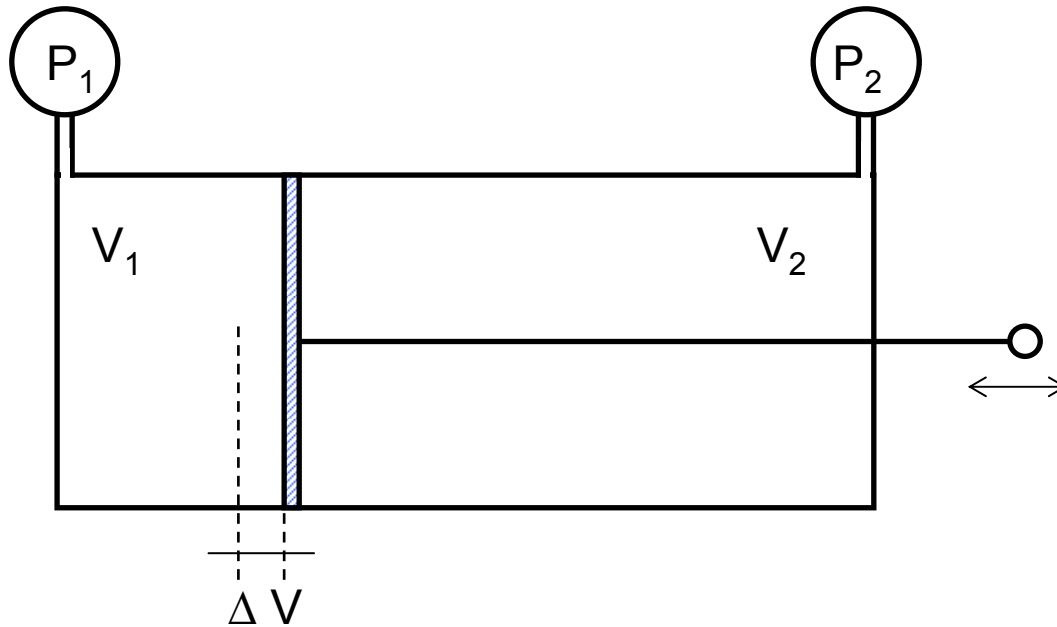
$$P.V = n.R.T$$

$$N.m^{-2}.m^3 = mol.J.mol^{-1}.K^{-1}.K$$

$$Nm = J$$

P	pressure	[Pa; N.m ⁻²]
V	volume	[m ³]
N	molecules	[mol]
R	gas constant	[J.mol ⁻¹ .K ⁻¹]
T	Temperature	[K]

In a closed volume, if temperature is constant: $PV = c$



Closed volume, assume temperature constant: $PV = c$

Moving the piston to the left...

$$P_1 \cdot V_1 = c_1 = P_1' (V_1 - \Delta V) \quad \text{and} \quad P_2 \cdot V_2 = c_2 = P_2' (V_2 + \Delta V)$$

$$P_1 \cdot V_1 = c = P_1' (V_1 - \Delta V)$$

$$P_1 \cdot V_1 = P_1' \cdot V_1 - P_1' \cdot \Delta V$$

$$V_1 (P_1 - P_1') = -P_1' \cdot \Delta V$$

$$V_1 = \frac{P_1' \cdot \Delta V}{(P_1' - P_1)}$$

$$P_2 \cdot V_2 = c = P_2' (V_2 + \Delta V)$$

$$P_2 \cdot V_2 = P_2' \cdot V_2 + P_2' \cdot \Delta V$$

$$V_2 (P_2 - P_2') = P_2' \cdot \Delta V$$

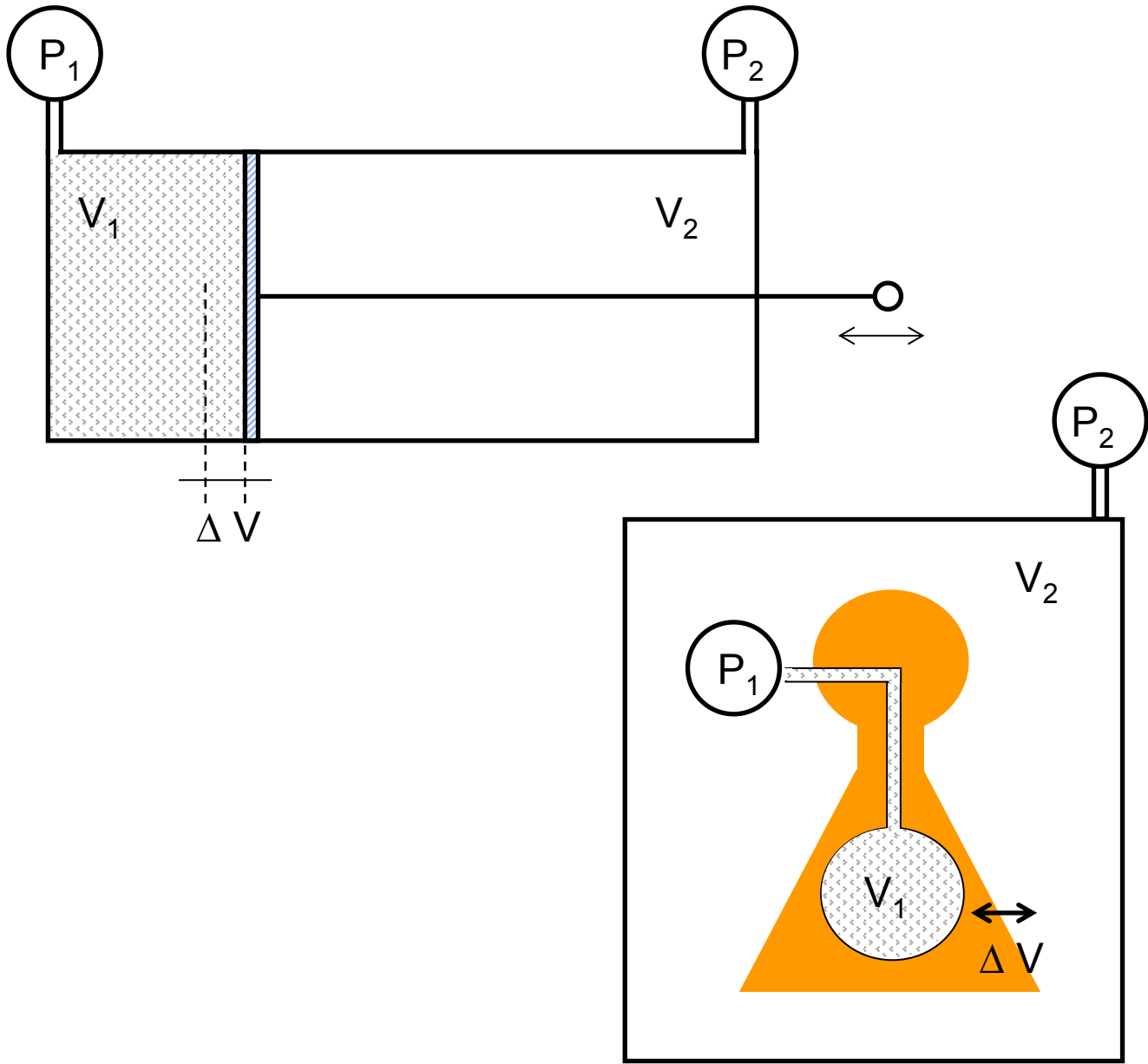
$$\Delta V = \frac{(P_2 - P_2')}{P_2'} V_2$$

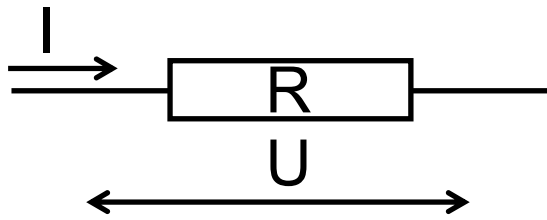
$$V_1 = \frac{P_1' \cdot \Delta V}{(P_1' - P_1)} \quad \Delta V = \frac{(P_2 - P_2')}{P_2'} V_2$$

$$V_1 = \frac{P_1'}{(P_1' - P_1)} \frac{(P_2 - P_2')}{P_2'} V_2$$

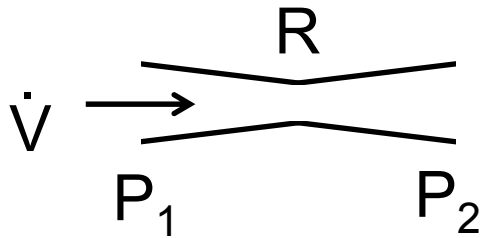
$$P_1' \approx P_2'$$

$$V_1 \approx \frac{\Delta P_2}{\Delta P_1} V_2$$





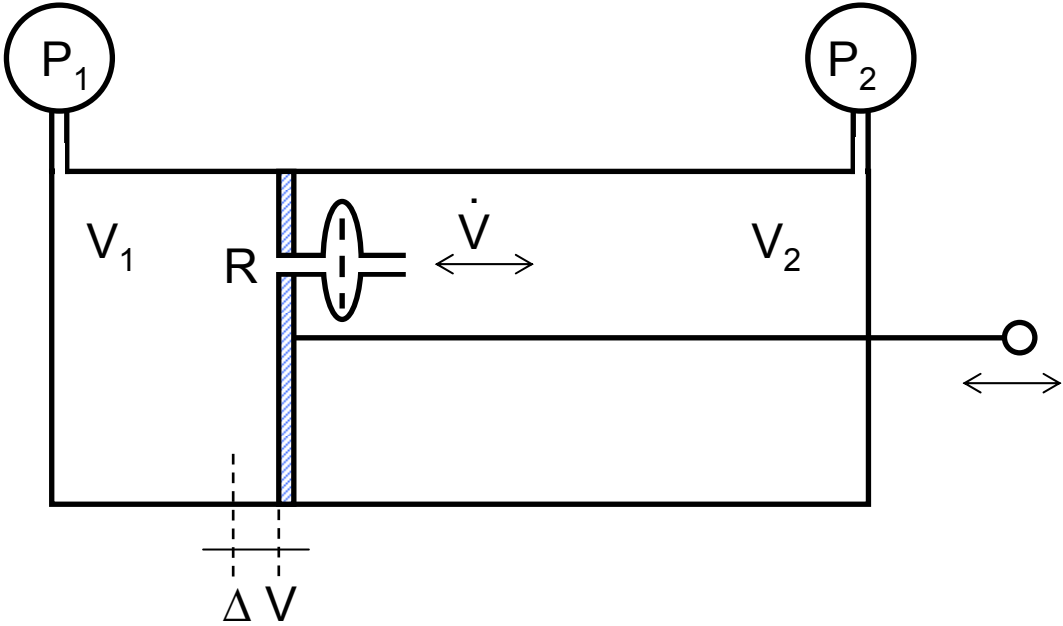
$$R = \frac{U}{I}$$



$$R = \frac{\Delta P}{\dot{V}}$$

$$R = \frac{P_1 - P_2}{\dot{V}} \quad \frac{\text{kPa}}{\text{l/s}}$$

Resistance



Resistance:

P_1 can not be measured, so we make the assumption:

$$\frac{V_1}{V_2} = \frac{P_2}{P_1} \quad \text{or:} \quad P_1 = P_2 \frac{V_2}{V_1}$$

$$R = \frac{P_1 - P_2}{\dot{V}} \quad \text{will be:}$$

$$R = \frac{P_2 \frac{V_2}{V_1} - P_2}{\dot{V}}$$

Resistance

$$R = \frac{P_2 \left(\frac{V_2}{V_1} - 1 \right)}{\dot{V}}$$

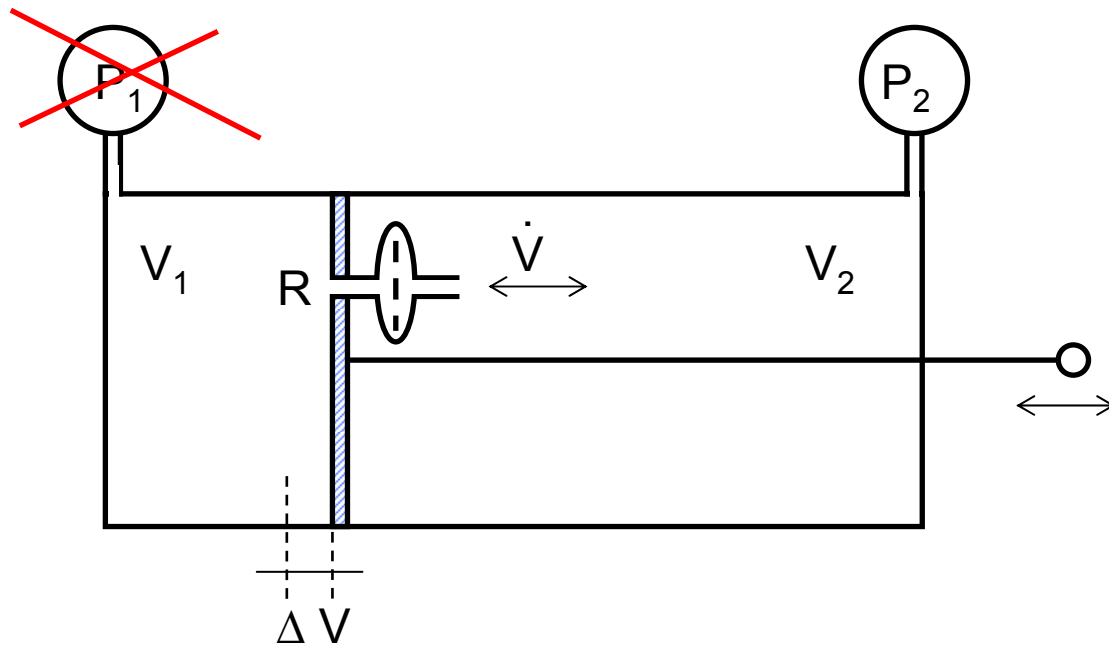
$$V_2 \gg V_1$$

$$R \approx \frac{P_2}{\dot{V}} \frac{V_2}{V_1} \quad \frac{\text{kPa}}{\text{l/s}}$$

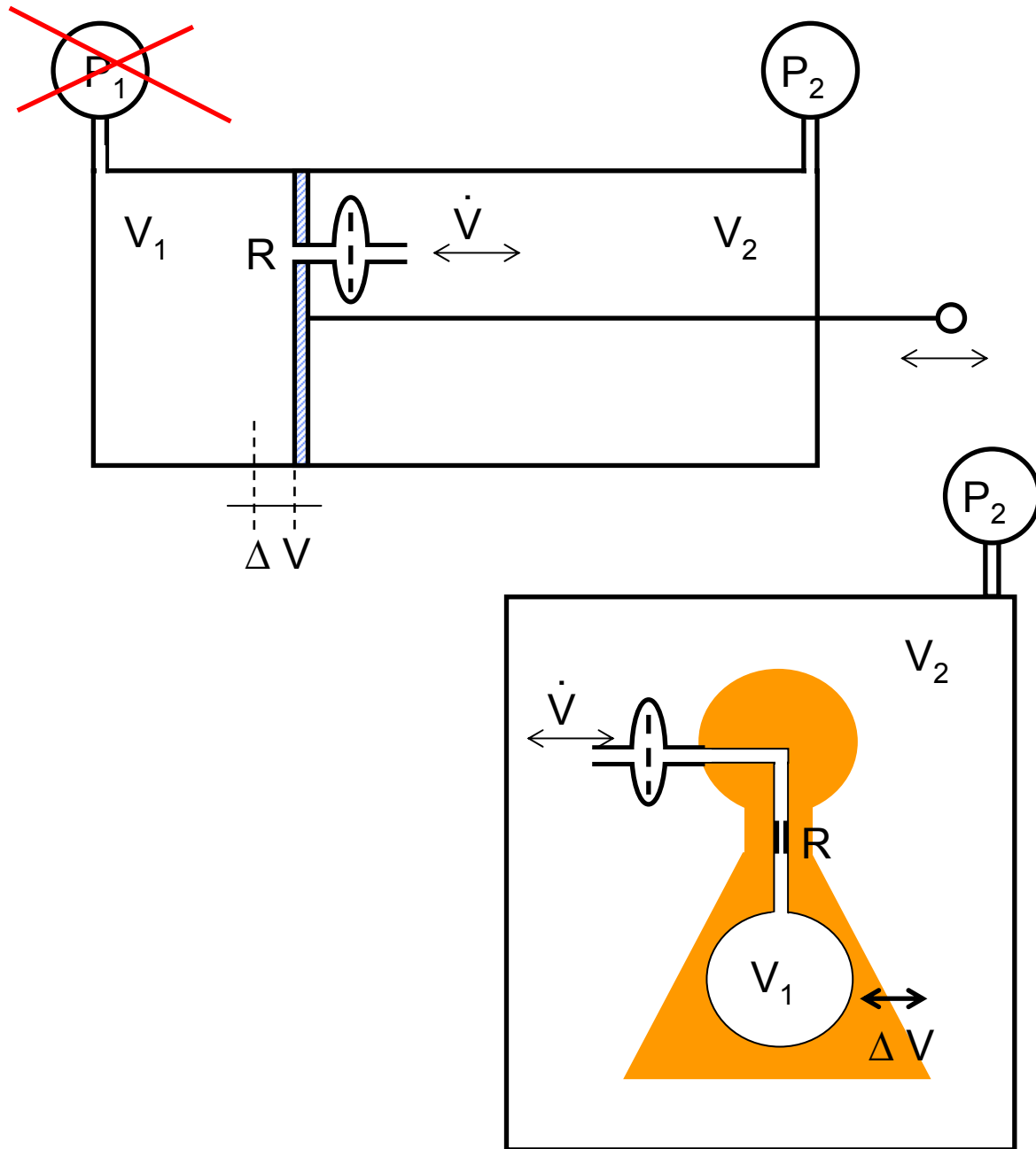
$\frac{V_2}{V_1}$ is a scaling factor

Resistance

$$R \approx \frac{P_2}{\dot{V}} \frac{V_2}{V_1}$$



Resistance



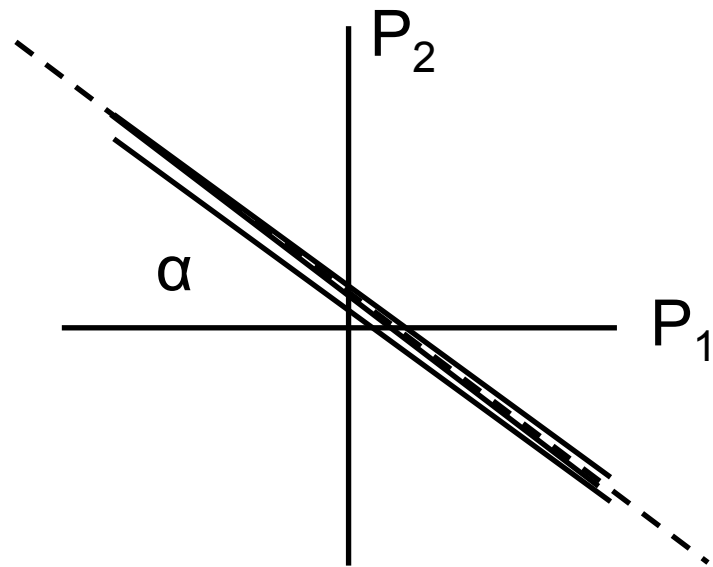
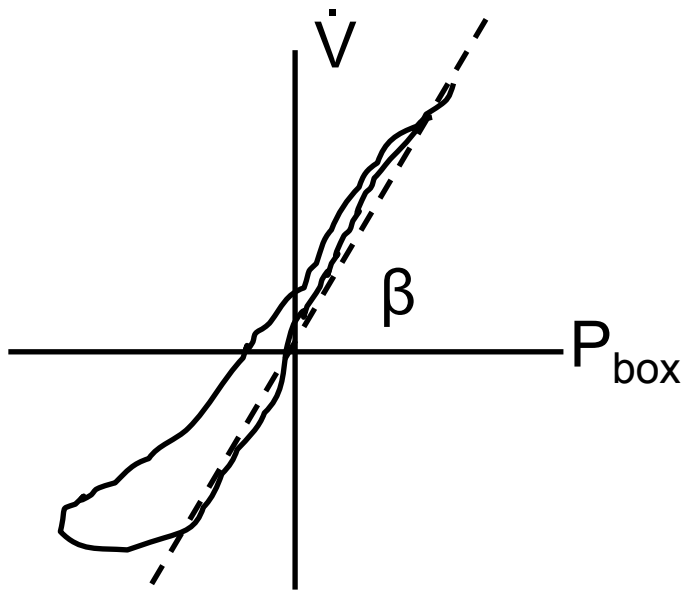
Resistance

$$R = \frac{P_2}{\dot{V}} \quad \frac{V_2}{V_1} \quad \frac{V_2}{V_1} = \frac{P_1}{P_2}$$

$$\frac{P_2}{\dot{V}} = \cotg \beta$$

$$\frac{\text{kPa}}{\text{l/s}}$$

$$\frac{P_1}{P_2} = \cotg \alpha$$



Heating up of the bodybox

Elements heating up the box:

- Body mass/ temperature
- **IN**spiration

Elements cooling the box:

- Thermal flow, depending on T_{amb} and T_{box} , thermal isolation box

The higher the Bodybox temperature:

- the lesser the pressure builds up during inspiration
- the lesser the cooling down of an exhaled volume
- the more heat leaks away

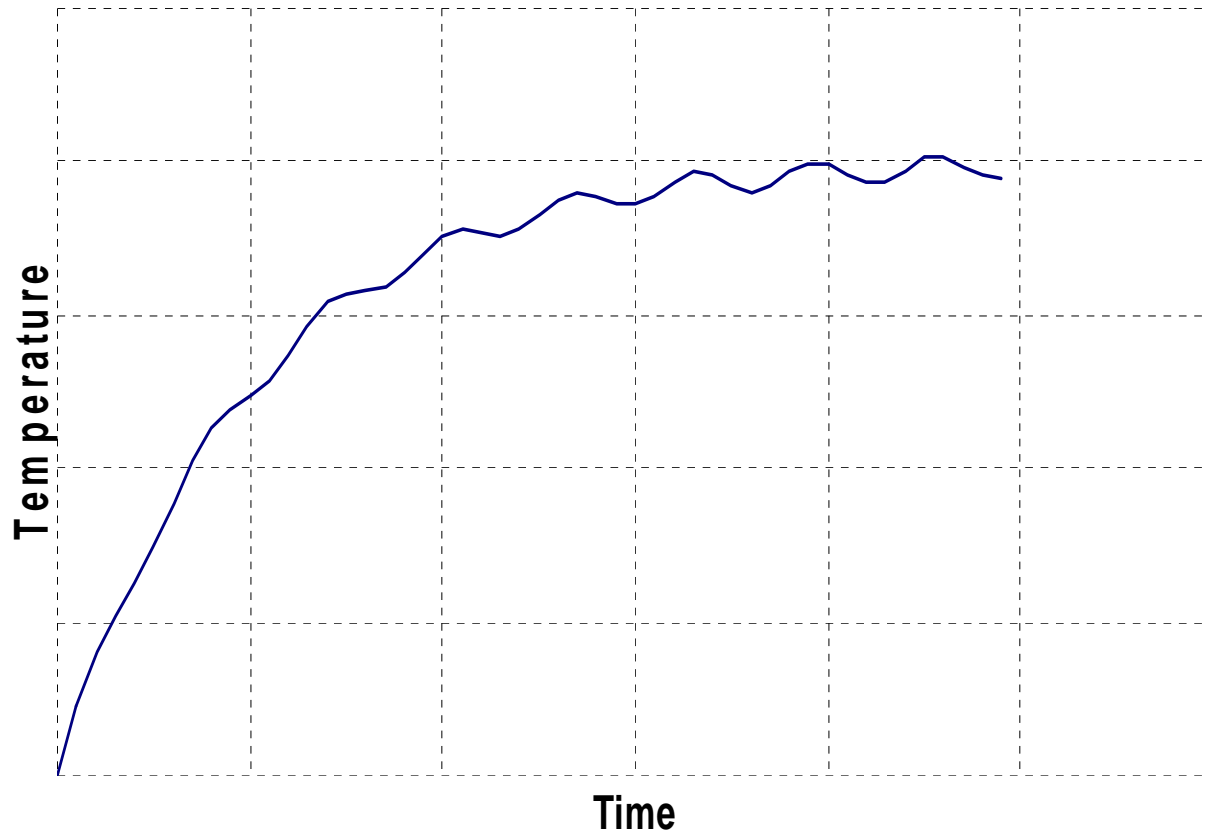
It becomes more thermally stable

Changing the temperature/ pressure in the box

- is more pronounced during inspiration

Every time we inspire, we cause a pressure artifact

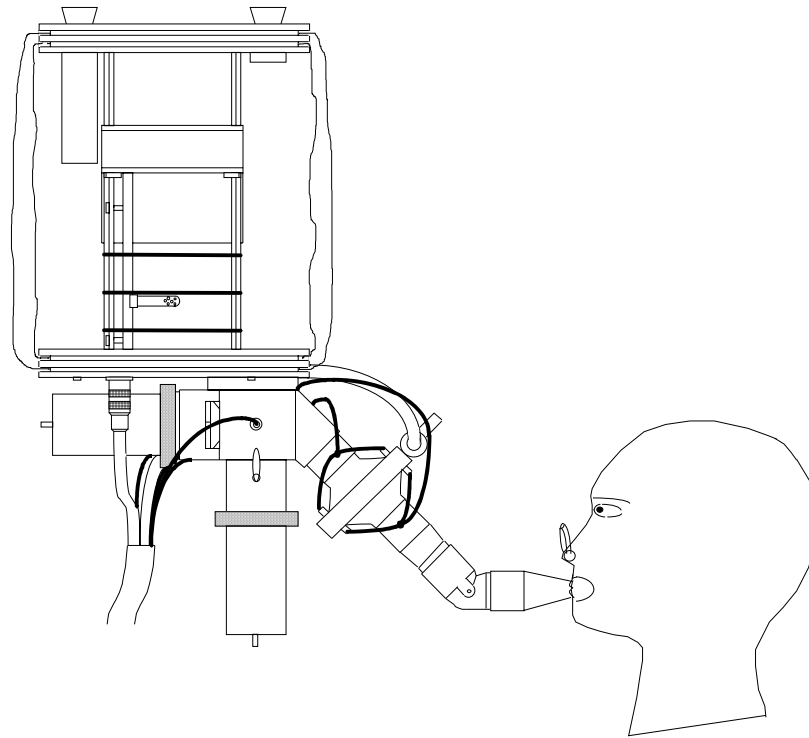
Heating up of the bodybox



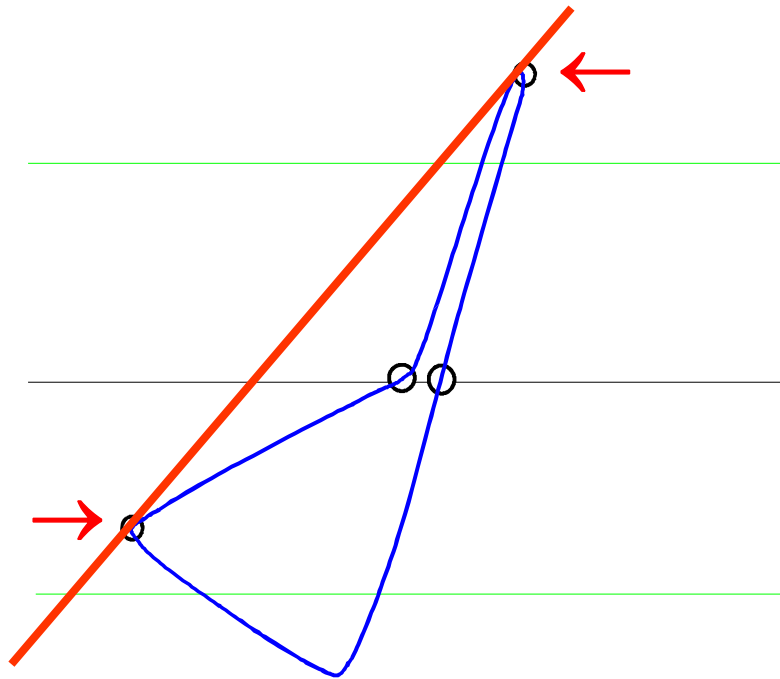
BTPS bag

No building up of pressure during inspiration

Today: corrected by software

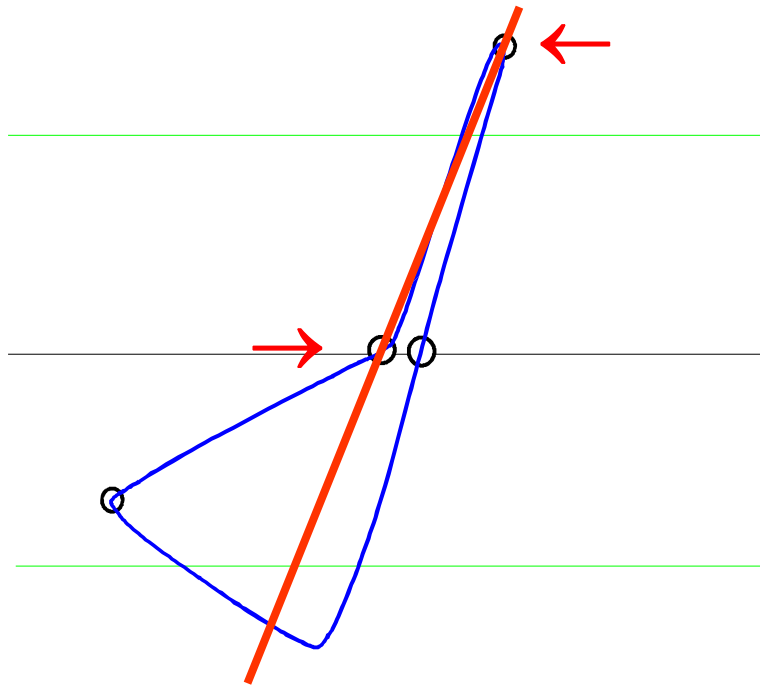


sR_{tot} (Ulmer)



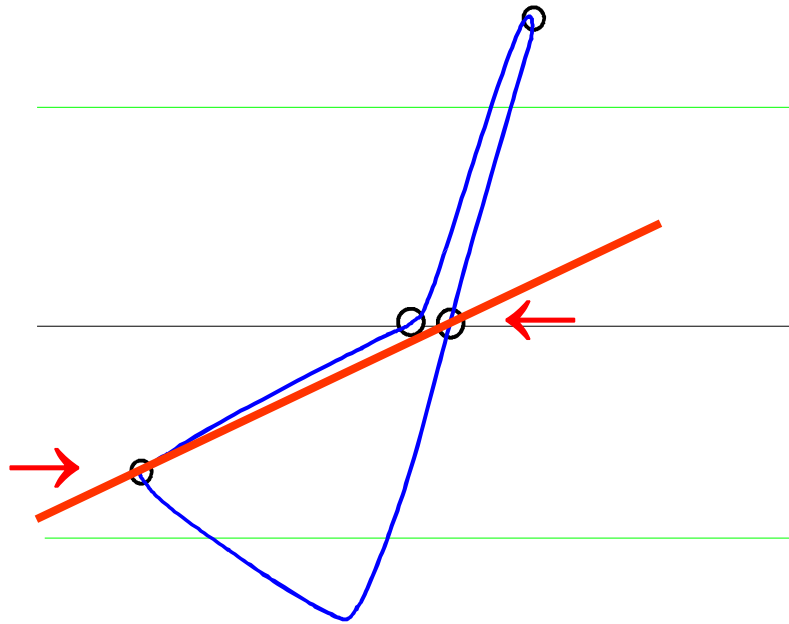
Connecting the maximal pressure points

sR_{in} (Ulmer)



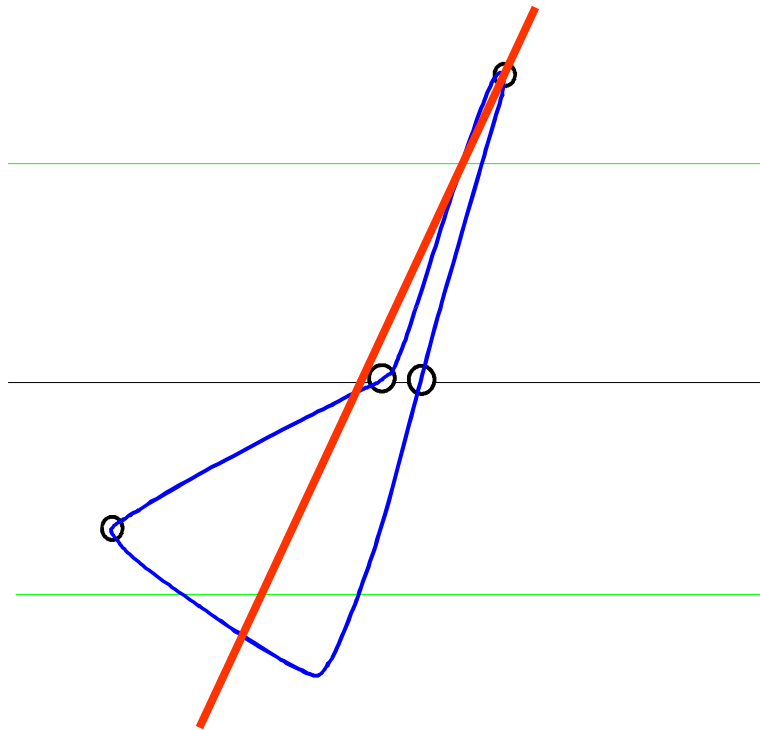
Connecting the maximal
pressure points
(inspiration)

sR_{ex} (Ulmer)



Connecting the maximal
pressure points
(expiration)

sR_{eff} (Matthys)



Area below the work of breathing-loop compared with the area below the Flow/Volume-loop (\approx least square fit)

$$\text{Raw} = \frac{\text{Work of breathing}}{\text{Area FV curve}}$$

Work of breathing = Raw . Area FV curve

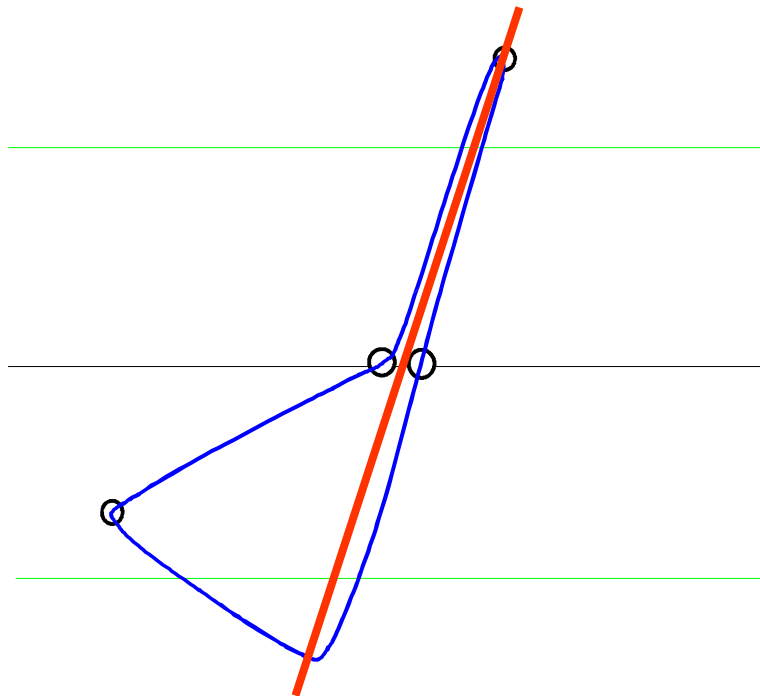
$$\text{WOB} = \frac{\text{N}}{\text{m}^2} \frac{\text{s}}{\text{L}} \frac{\text{L}}{\text{s}} \text{L}$$

$$\text{WOB} = \frac{\text{N}}{\text{m}^2} \text{m}^3$$

WOB = pressure x volume

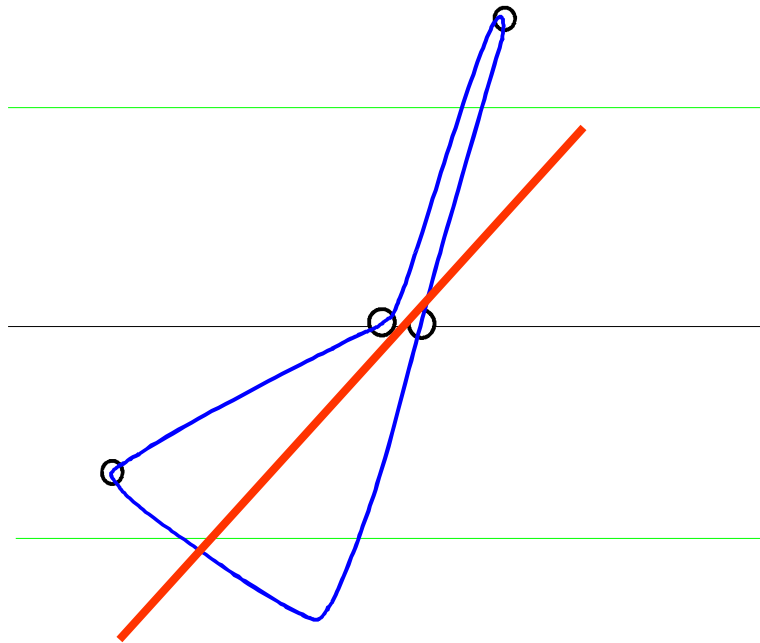
$$\text{Raw} = \frac{\text{pressure}}{\text{flow}} \times \frac{\text{volume}}{\text{volume}}$$

$sR_{\text{eff,in}}$ (Matthys)



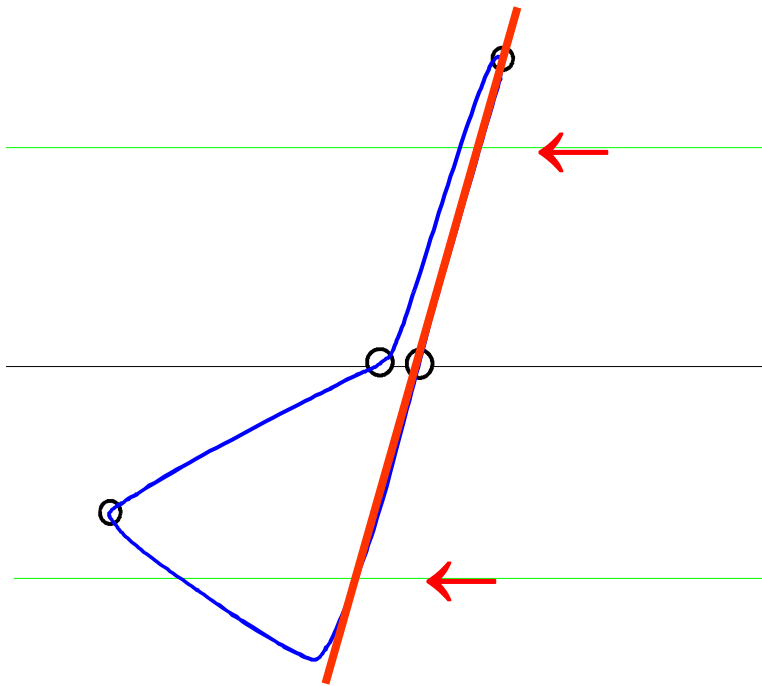
Area below the work of breathing-loop compared with the area below the Flow/Volume-loop during inspiration

$sR_{\text{eff,ex}}$ (Matthys)



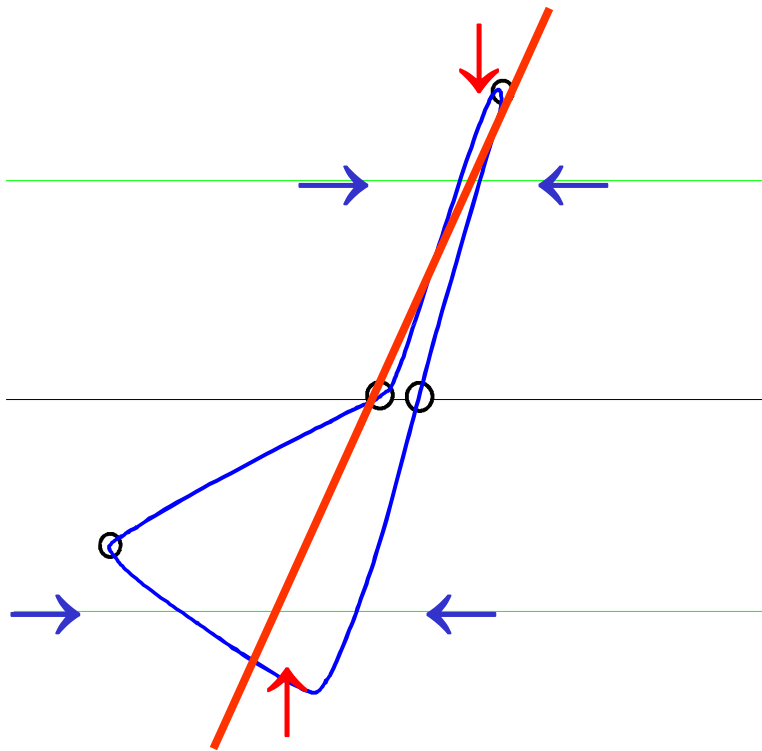
Area below the work of breathing-loop compared with the area below the Flow/Volume-loop during expiration

Resistance $sR_{0.5}$



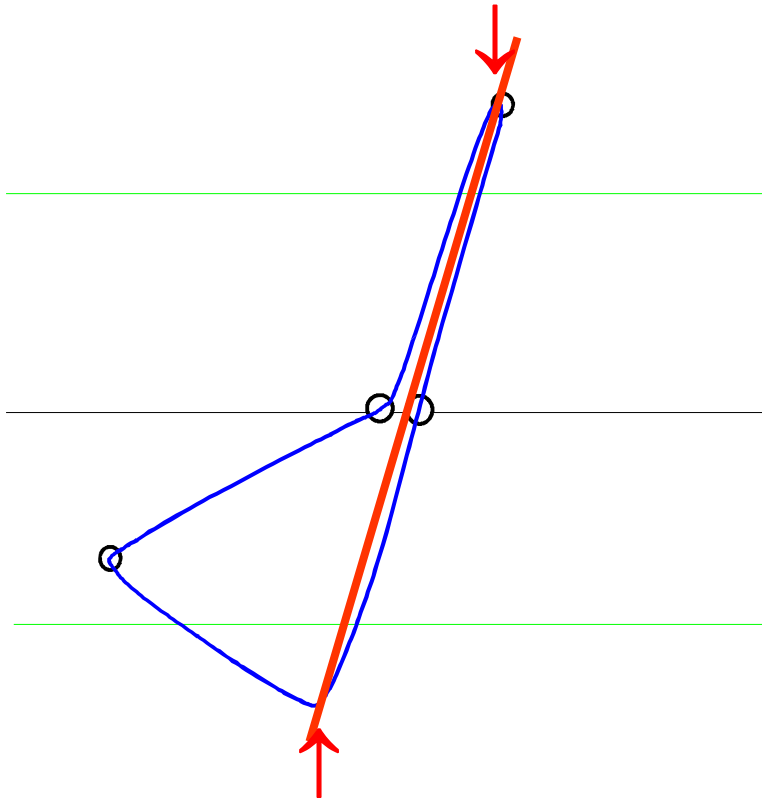
connecting the
flow-values at
+ 0.5 and - 0.5 L/sec

Resistance sR_{mid}



connecting the mean
flow-values at
+ 0.5 and - 0.5 L/sec

Resistance sR_{peak}

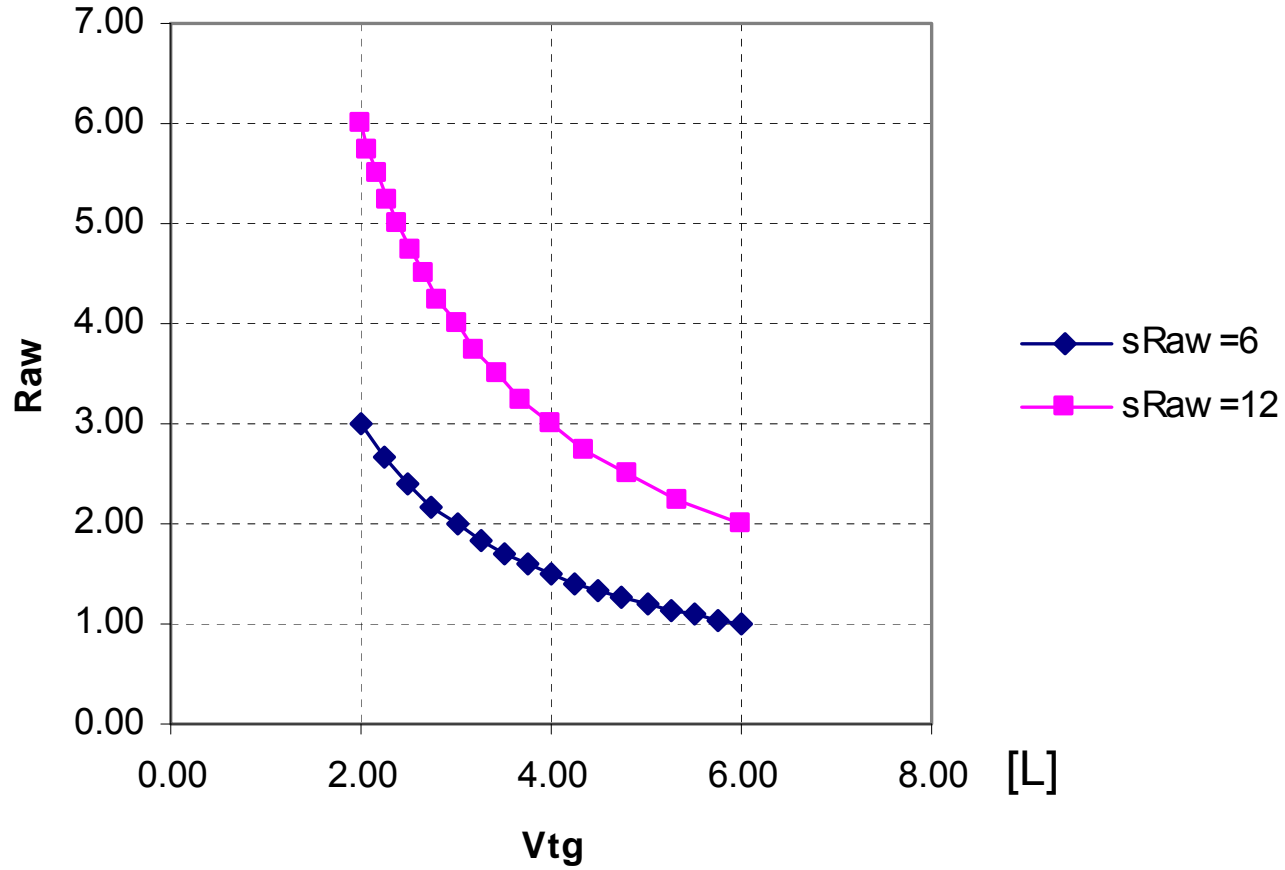


connecting the maximal
and minimal flow-values

Resistance: Raw vs sRaw

$$Raw=f(Vtg)$$

[kPa.s.L⁻¹]



Resistance: R_{aw} vs sR_{aw}

Hyperbolic relation: $R_{aw} = f(V_{tg})$

$$R_{aw} \cdot V_{tg} = c$$

$$R_{aw} = \frac{c}{V_{tg}}$$

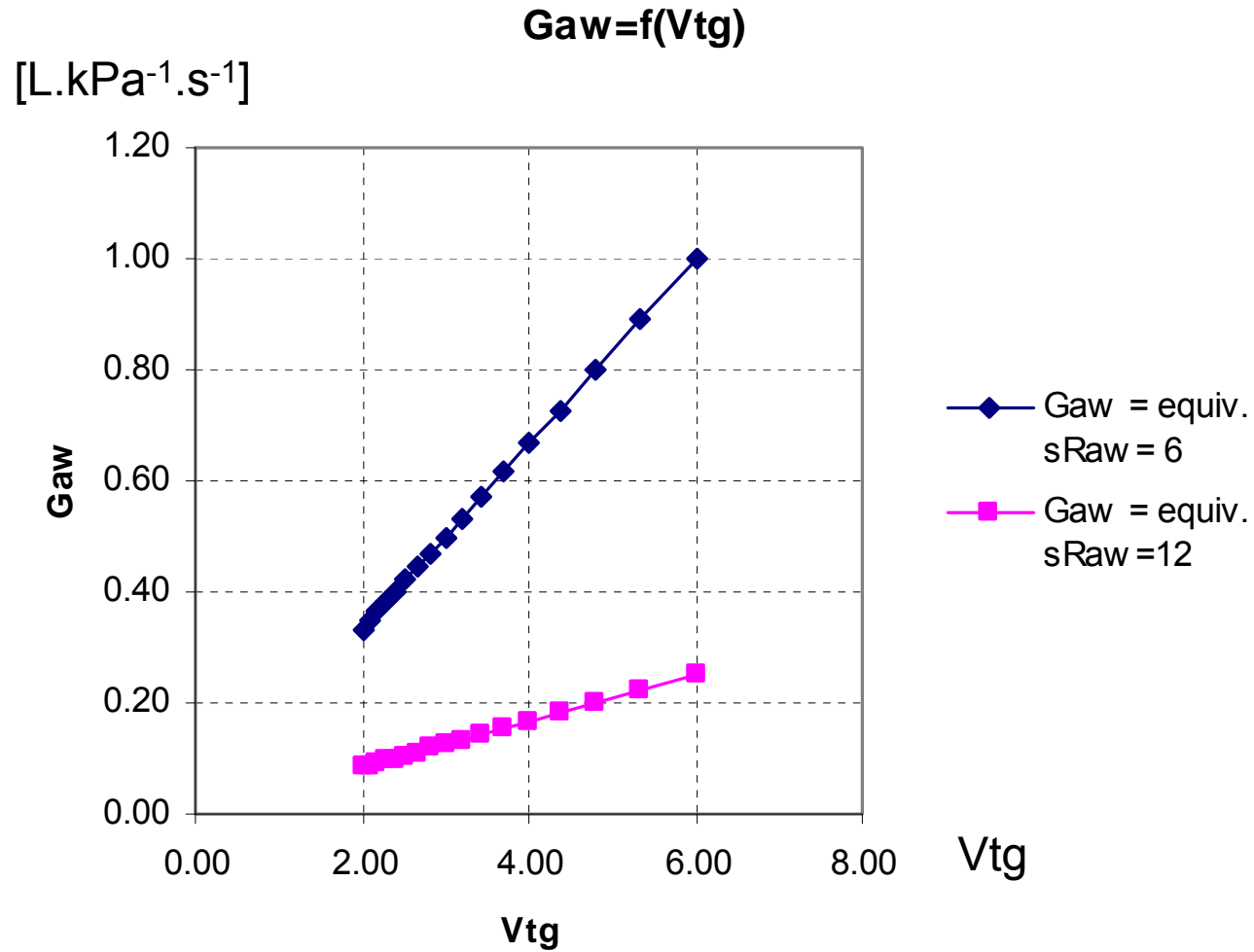
$$\frac{1}{R_{aw}} = \frac{1}{c} V_{tg}$$

$$G_{aw} = c' V_{tg}$$

note:

$$c' = \frac{1}{sR_{aw}} = \frac{1}{R_{aw} \cdot V_{tg}}$$

Resistance: Raw vs sRaw

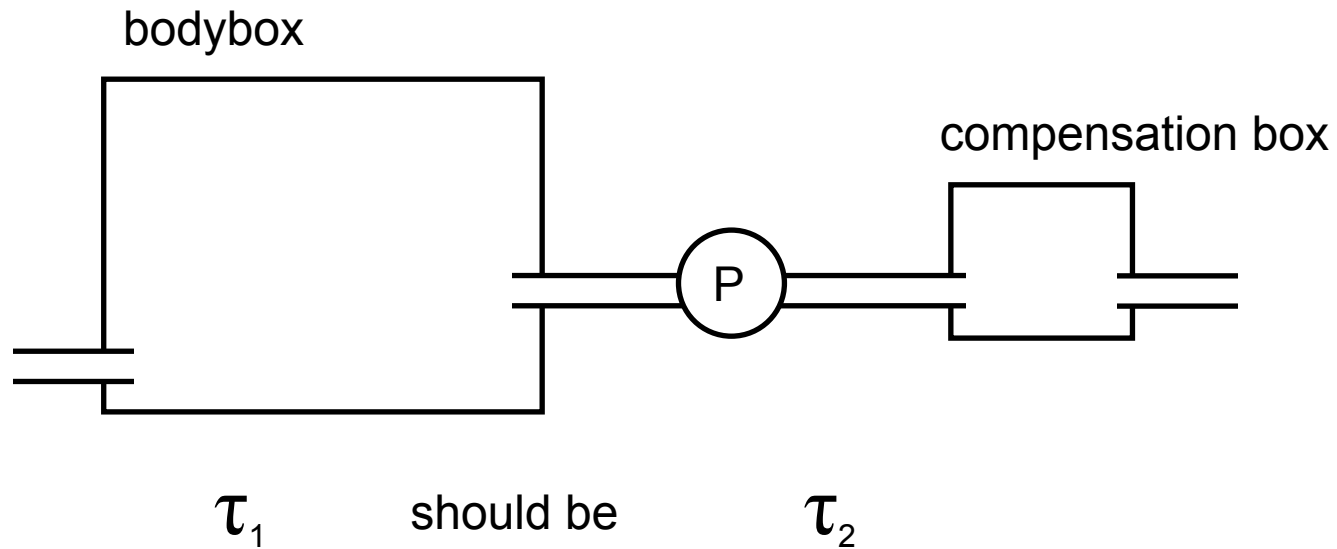


A bodybox needs a certain leak. If not: pressure would go up too much due to thermal effect

We hence deliberately make a leak of about 4...7 seconds time constant.

That is why the sinusoidal calibration pump frequency should match the breathing frequency for correct calibration. If the frequency is different, the dynamic response is not identical.

A bodybox needs a certain leak. If not: pressure would go up too much due to thermal effect



If time constants not equal, enormous pressure artefacts

Calibrating the pressure transducer of the box:

$$P_2 \cdot V_2 = c = P_2' (V_2 + \Delta V)$$

$$P_2 \cdot V_2 = P_2' V_2 + P_2' \Delta V$$

$$V_2 (P_2 - P_2') = P_2' \Delta V$$

$$(P_2 - P_2') = \frac{P_2' \Delta V}{V_2}$$

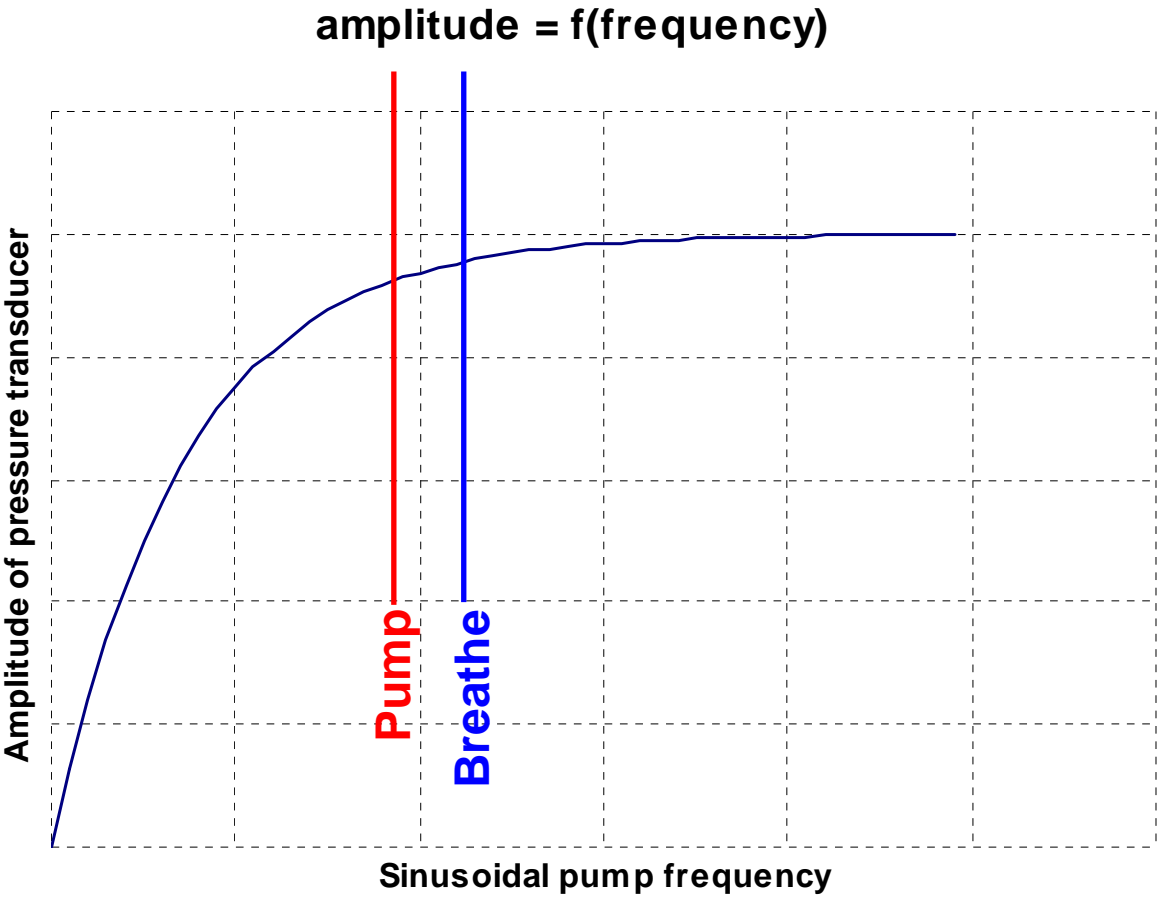
ΔV : calibration pump
volume

V_2 : volume of the box

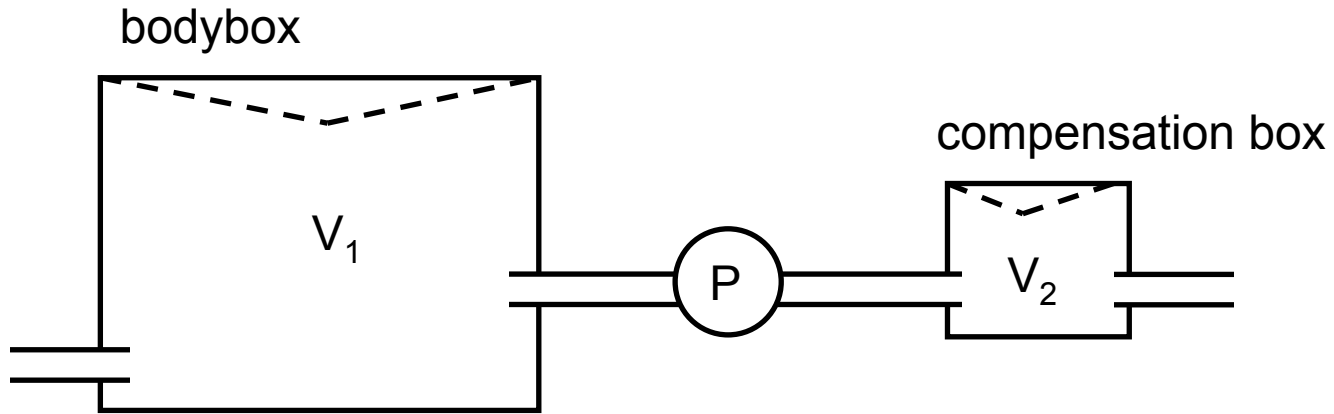
$P_2' - P_2$: amplitude of the
pressure transducer

This is only valid in a box which has no leak!!

Calibrating the pressure transducer of the box:



Pressure effect: deformation of box and compensation box



$$P_1 \cdot V_1 = C_1$$

$$P_2 \cdot V_2 = C_2$$

$$P_1' = \frac{C_1}{V_1 - \Delta V_1}$$

$$P_2' = \frac{C_2}{V_2 - \Delta V_2}$$

Pressure effect: deformation of box and compensation box

$$P_1' \approx P_2'$$

$$\frac{C_1}{V_1 - \Delta V_1} = \frac{C_2}{V_2 - \Delta V_2}$$

$$\frac{P_1 \cdot V_1}{V_1 - \Delta V_1} = \frac{P_2 \cdot V_2}{V_2 - \Delta V_2}$$

$$P_1 \approx P_2$$

$$\frac{V_1}{V_1 - \Delta V_1} = \frac{V_2}{V_2 - \Delta V_2}$$

Pressure effect: deformation of box and compensation box

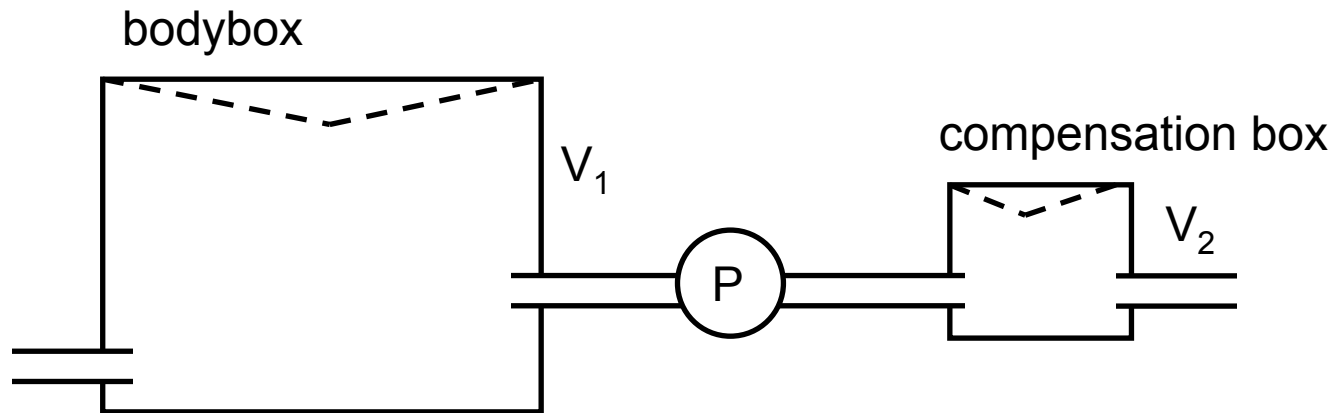
$$V_1 (V_2 - \Delta V_2) = V_2 (V_1 - \Delta V_1)$$

$$V_1 V_2 - \Delta V_2 V_1 = V_2 V_1 - \Delta V_1 V_2$$

$$- \Delta V_2 V_1 = - \Delta V_1 V_2$$

$$\Delta V_2 = \frac{\Delta V_1 V_2}{V_1}$$

Pressure effect: deformation of box and compensation box



$$\Delta V_2 \cdot V_1 = \Delta V_1 \cdot V_2$$

ΔV depends on both material properties and mechanical construction